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suppose that 5.6 mol of aluminum reacts with 4.4 mol of bromine. 1.

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Q. What is the percent yield if 0.856 g of  $\text{NH}_3$  is actually obtained in the lab during the following reaction:  $4\text{NH}_3 + 5\text{O}_2 \rightarrow 4\text{NO} + 6\text{H}_2\text{O}$  How many grams of NO are formed if 6.30g of ammonia react with 1.80g of oxygen?

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